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The crystal structures of Nb₃Au and V₃Au. By ELIZABETH A. WOOD and BERND T. MATTHIAS, Bell Telephone Laboratories, Inc., Murray Hill, N. J., U.S. A.

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Recently Geller, Matthias & Goldstein (1955) described several new intermetallic compounds with the ' β -wolfram' structure (A15) (*Strukturbericht*, 1937) discovered in the search for superconductors with high transition temperatures. Two new compounds have now been added to the list: Nb₃Au and V₃Au. Nb₃Au becomes superconducting at 11.5° K. and V₃Au is not superconducting above 1.1° K. One other gold compound has been reported to have the β -wolfram structure. This is Ti₃Au, reported by Duwez & Jordan (1952).

The compounds Nb₃Au and V₃Au resulted from the reaction of stoichiometric quantities of niobium powder and gold wire and of massive vanadium and gold wire in a helium arc furnace. The temperature of the reaction is not known, but the sample was certainly heated to above 2000° C. Each charge was fused three times to promote homogeneity.

X-ray diffraction powder photographs were taken of

 Table 1. Comparison of calculated with observed intensities, visually estimated

(vs = very strong; s = strong; m = medium; w = weak)

	Nb ₃ Au		V ₃ Au	
hkl	Obs.	Calc.(ord.)	Obs.	Calc.(ord.)
110	m	25	m-s	47
200	8	46	m	27
210	s–vs	56	\boldsymbol{w}	12
211	vs	100	8	60
220	vw	4	vw	7
310	w	6	vw	10
222	vvw	0.8	Abs.	0.4
320	m	11	Abs.	2
321	8	49	m	29
400	m	13	vw	6
411, 330	vw-w	4	vw	6
420	m	14	vw	9
421	m	10	Abs.	2
332	m	12	vw	8
422	vvw	2	vvw	3
510, 431	w	8	vw	11
520, 432	m	11	vvw	3
521	m-s	21	w	17
440	m	14	vw	8
530, 433	vvw	3	vw	8
600, 442	m	16	w	16
610	w	5	Abs.	2
611, 532	m - s	40	8	52

Note.—The two films were not comparable with each other in intensity; the 'observed' intensities are therefore on different scales.

Compound	a ₀ (Å)	Interatomic distances (Å)			
		B-A	$(A-A)_1$	$(A-A)_2$	
$Nb_{3}Au$	$5 \cdot 21 \pm 0 \cdot 01$	2.91	2.61	3.19	
V ₃ Ău	4.88 ± 0.01	2.73	$2 \cdot 44$	2.99	

Each B atom is surrounded by 12 A atoms at a distance B-A; each A atom is at distance $(A-A)_1$ from an A atom on either side of it, $(A-A)_2$ from eight other A atoms, and B-A from four B atoms which surround it in a distorted tetrahedron.

the resulting product, using a Norelco camera of 114.6 mm. diameter and Cu K radiation. The patterns were indexed on a primitive cubic cell with the lattice constants given in Table 2. Reflections of the type (hhl) with l odd are absent. In addition, reflections (410), (430), (531), (621) and (540) are absent. The possibility that compounds with the β -wolfram structure would result from the fusion of these elements in the appropriate proportions had been anticipated, and the close similarity between these powder patterns and those of the compounds reported by Geller et al., indicated that they are isostructural. The absences are those appropriate to the β -wolfram structure. This structure belongs to space group $O_h^3 - Pm3n$ with two atoms of one type in position 2(a): (0, 0, 0), $(\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$ and six atoms of the other type in positions 6(c): $\pm(\frac{1}{4}, 0, \frac{1}{2}; \bigcirc)$. Alternatively, these six atoms may be placed in 6(d): $\pm(\frac{1}{4}, \frac{1}{2}, 0; \bigcirc)$ and the resulting structure would be the same. The twofold and sixfold positions of space group T_d^4 -P43n are the same as those of space group $O_h^3 - Pm3n$. The higher-symmetry space group is chosen.

Relative intensities for each of these compounds were computed on the basis of the above structure, using the formula

$$I \propto p |F_{hkl}|^2 rac{1 + \cos^2 2 heta}{\sin^2 heta \, \cos heta} imes 10^{-5}$$
 ,

where p is the multiplicity factor, F_{hkl} the structure amplitude, and the remaining term is twice the combined Lorentz and polarization factors. James & Brindley's (1931) scattering factors were used for gold and vanadium, and the Thomas-Fermi factors from the International Tables (1935) for niobium.

The results are shown in Table 1, compared with the observed intensities, visually estimated. The agreement is so close that the β -wolfram structure is indicated both for V₃Au and for Nb₃Au. The fact that the (110) reflection in both cases is somewhat weaker than indicated by calculation is interpreted as due to absorption.

In each pattern two 'very very weak' extra lines were observed. The *d* values for those in the Nb₃Au pattern were 2.42 and 2.24 Å; for those in the V₃Au pattern, 2.28 and 1.44 Å. These lines are not the strongest lines for any of the following substances: Nb, Au, V, Nb₂O₅, V₂O₃, V₂O₅. Their origin is unexplained.

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